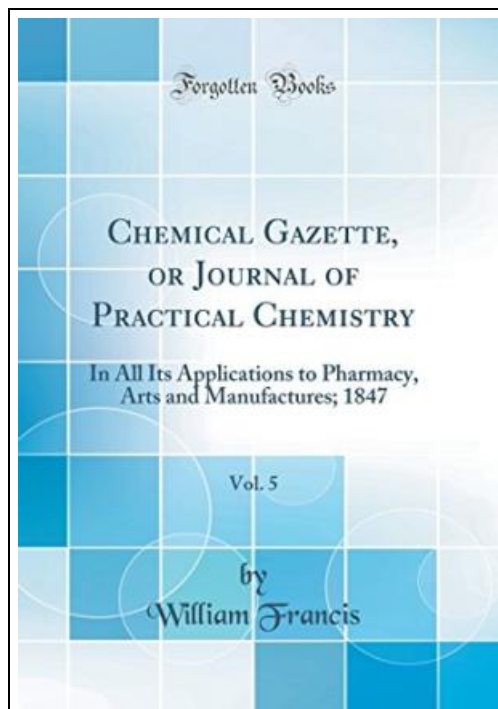


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Forgotten Books, 2017. Hardback. Condition: New. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.Excerpt from Chemical Gazette, or Journal of Practical Chemistry, Vol. 5: In All Its Applications to Pharmacy, Arts and Manufactures; 1847 On the Precipitation and Separation of Alumina and Peroxide of Iron. By Dr. W. Knop. In the precipitation Of the peroxide of iron and alumina by ammonia, and their subsequent separation by potash, it is a matter of common experience that these two oxides are extremely difficult to obtain perfectly free from one another. It was to be expected that a more perfect separation would be obtained if this mutual affinity were destroyed previously by combining one of the two substances with some other body. When the precipitate Of hydrated oxide of iron and alumina is boiled with potash with the addition of a suitable quantity of sulphuret of ammonium, until the liquid, which at first appears green, deposits, after being left quiet for a time, some black fakes, and appears yellowish, owing to the formation Of a slight excess of sulphuret Of potassium, and no longer passes green through the filter, which is very soon attained, the alumina, on subsequent precipitation, is Obtained on slow desiccation as a transparent mass, and on quick drying and calcining of such a pure white colour as to leave no doubt of its being extremely pure. If we add to this the observations recently published by Malaguti and Durocher\*, with respect to the complete precipitation of alumina, which was found to be most perfect under all circumstances with sulphuret of ammonium, it is evident that these two oxides should, when it is practicable, be directly precipitated with sulphuret of ammonium, the precipitate washed with water containing a little sulphuret of ammonium,...

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