

Blaue Mahrchen Fur Alte Und Junge Kinder



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Reviews

The ideal publication i ever read through. It is writter in simple words and never hard to understand. Your daily life span is going to be convert once you full looking over this ebook.
(Tanner Willms PhD)

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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 38 pages. Original publisher: Aberdeen Proving Ground, MD : Army Research Laboratory, 2010 OCLC Number: (OCoLC)710812456 Subject: Nitric acid. Excerpt: . . . Appendix. Mechanism Data Tables The monomethylhydrazine-red fuming nitric acid (MMH-RFNA) mechanism discussed in this report comprises thermochemical parameters and rate expressions for 81 species and 513 elementary reactions. Thermochemical parameters are given in table A-1, and the rate expressions are given in table A-2. The thermochemical parameters include (1) heat of formation at 298 K $H^{\circ}(298)$, (2) entropy at 298 K $S^{\circ}(298)$, and (3) the constant pressure heat capacity $C_p(T)$ as a function of temperature T . For reactions whose rates (k) are not a function of pressure, the Arrhenius expression $k = A \exp(-E/RT)$ calculates the rate at a specified temperature from the given frequency factor (A), temperature exponent (b), activation energy (E), and the ideal gas constant (R). For reactions involving a generalized collider species (M) (and whose rates are therefore pressure dependent), an effective collider concentration (C) is calculated from the expression $C = M p / (RT)^X$, where p is the pressure, X is the mole fraction, and i is the collider efficiency of species i . Collider efficiencies are assumed to be 1.0 unless otherwise specified. Then, one of three expressions is employed for the rate calculation. If T and $H^{\circ}(S)$ is specified, the 1 Tsang and Herron form is used and constants for that form are given,...



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